product was extracted twice with an equal vol. of C_6H_6 and dried over anhydrous Na_2SO_4 . GLC analysis (OV-101, 100°, isothermal program) indicated 95% purity. The MS m/e 138 (M⁺) and UV spectrum λ_{max} 368 nm [14] were consistent with MQ.

Enzymatic reaction. Reaction mixtures (50 ml) contained 10mM sodium phosphate, pH 6.5, 10 mM $\text{ H}_2\text{O}_2$ and various cones of phenolic substrates as indicated. Reactions were started by the addition of 50 µl peroxidase (50 µg) and were carried out at 38°. At the end of the incubation, 2 ml 2 N H_2SO_4 were added and the mixture was immediately extracted with EtOAc (2 × 30 ml). The combined organic fractions were dried and evapd to dryness in N_2 . Trimethylsilanation of enzymatic products and standards was carried out by adding bis-{N,O-trimethylsilyl}trifluoroacetamide-pyridine (1:1) to the dry residue and heating at 80° for 5 min.

Gas chromatography. GLC was carried out with a Varian Model 1700 instrument fitted with a glass column (180 \times 0.2 cm i.d.) packed with 3% OV-101 on chromosorb Q 100/120 (Applied Science). The oven temp. was programmed from 100 to 270° at 10°/min unless indicated otherwise. MS was carried out with a Dupont Model 21-491B equipped with the same instrument and column for GLC. The spectra were obtained at 70 eV.

Acknowledgements—This research was supported by the Crown Zellerbach, International Paper and Weyerhaueser Companies and by the Gottesman Foundation.

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0031-9422/79/1201-2022 \$02/00/0

OVALIN, A NEW PIPECOLIC ACID FROM MILLETIA OVALIFOLIA SEEDS

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(Received 14 May 1979)

Key Word Index-Milletia ovalifolia; Leguminosae; ovalin.

In continuation of our earlier work [1], further examination of Milletia ovalifolia seeds led to the isolation of a new pipecolic acid, ovalin, from the methanol extract. Ovalin was obtained chromatographically pure as colourless needles from EtOH-Me₂CO; mp 280-281° (decomp.); R_f 0.36 (PC n-BuOH-HCO₂H-H₂O, 15: 3:2). It analysed for $C_7H_{13}NO_3$; M^+ 159; $[\alpha]_D^{20}$ -41.2° (H₂O) was chemically neutral and insoluble in all organic solvents except alcohol in which it was partially soluble, but was readily soluble in H₂O. There was no UV absorption in the range 200-300 nm. The IR spectrum (KBr) indicated an —OH group (ν 3350 cm⁻¹), —O of carboxylate (1630 cm⁻¹) and a quaternary ammonium group (=N⁺=) (1400 cm⁻¹). It did not

respond to tests for amino acids, peptides and alkaloids. The inertness towards most of the reagents and the presence of nitrogen suggested it to be a pipecolic acid derivative, some of which are known to occur in the seeds of leguminous plants [2].

The ¹H NMR spectrum (60 MHz, solvent D_2O , values) indicated the presence of N—Me group, methine and methylene protons. The spectrum closely resembled that of 4-hydroxypipecolic acid recorded in D_2O [3–5]. A quintet centred at 4.3 ($J_{ae} = J_{ee} = 3.5$ Hz, 1H) was assigned to an equatorial C-4 proton. A pair of doublets centred at 3.85 ($J_{ae} = 5$; $J_{aa} = 11$ Hz, 1H) was assigned to an axial C-2 proton because of the large coupling constant, characteristic of axial-axial neigh-

bours. The C-6 protons and four methylene protons appeared as multiplets at 3.2-3.6 (2H) and 1.9-2.4 (4H), respectively. A singlet at 3.0 (3H) was attributed to the =N-Me protons.

Ovalin formed a methyl ester with ethereal CH_2N_2 as a gummy mass, R_f 0.58 (PC n-BuOH-HCO₂H-H₂O, 15:3:2); IR (KBr) cm⁻¹: 3400 (OH) and 1730 (C=O of ester). ¹H NMR (60 MHz, solvent D₂O): δ 4.1-4.6 (2H, m), 3.55-3.85 (2H, m), 3.35 (3H, s), 3.3 (3H, s) and 2.0-2.5 (4H, br m). On acetylation with Ac₂O-HClO₄, ovalin formed a monoacetate as an oil; R_f 0.57 (PC n-BuOH-HCO₂H-H₂O; 15:3:2). IR (Nujol) cm⁻¹: 1750 and NMR (solvent D₂O); δ 4.2 (2H, br m), 3.5 (2H, m), 3.0 (3H, s, =N-Me), 2.3 (3H, s, —OAc) and 1.9-2.3 (4H, m). Formation of a monoacetate and methyl ester supported the presence of a hydroxyl and carboxyl group. Based on the above data, structure (1) is proposed for ovalin which is supported by MS fragmentations m/e 114 (99.5) (M⁺ - 45), 96 (100) (M⁺ - 45 – 18) and 70 (99.9%).

Ovalin showed a positive Cotton effect at 205 nm ($\Delta \varepsilon + 0.00453$) in the circular dichroism spectrum recorded in H₂O similar to (-)-trans-4-hydroxypipecolic acid which has a positive Cotton effect at 204 nm ($\Delta \varepsilon + 0.0061$) in H₂O [6]. The above data suggest that ovalin is (2S,4R)-4-hydroxy-N-methylpipecolic acid.

$$\begin{array}{c}
OH \\
H
\\
\downarrow \\
N
\\
COOH
\end{array}$$

$$\begin{array}{c}
HO \\
H
\\
\downarrow \\
6
\\
N_1
\\
\downarrow \\
Me
\end{array}$$

$$\begin{array}{c}
OH \\
H
\\
\downarrow \\
Me
\end{array}$$

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Phytochemistry, 1979, Vol. 18, pp. 2022-2023. © Pergamon Press Ltd. Printed in England.

0031-9422/79/1201-2022 \$02.00/0

N-METHYLTYRAMINE, A BIOLOGICALLY ACTIVE AMINE IN ACACIA SEEDS

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(Received 25 May 1979)

Key Word Index—Acacia; Leguminosae; chemotaxonomy; N-methyltyramine; migraine.

Abstract—The seeds of Acacia species belonging to the 'pennata' group characteristically contain N-methyltyramine (approximately 0.5% dry weight). Like tyramine, N-methyltyramine increases blood pressure in the anaesthetized rat, relaxes guinea pig ileum and increases both the force and rate of contraction of guinea-pig right atrium by inducing the release of noradrenaline.

INTRODUCTION

Seven Acacia species, A. bonariensis Gill. ex Hook & Arn., A. brevispica Harms., A. caesia W. & A., A. kraussiana Meisn. ex Benth., A. schweinfurthii Brenan & Exell, A. pennata (L.) Willd., A. pentagona (Schumach & Thonn.) Hook f., belonging to the 'pennata' group of Bentham's series Vulgares were reported recently to accumulate an unidentified ninhydrin-reacting compound in their seeds [1]. We have now isolated this compound from seeds of A. schweinfurthii Brenan & Exell, and identified

it as N-methyltyramine (NMT).

This is, as far as we are aware, the first report of NMT in Acacia seeds but it is known to occur in leaves of A. rigidula, A. roemeriana and A. berlandieri [2, 3]. The leaves of A. berlandieri (Guajillo) also contain tyramine, and when eaten by sheep and goats in the Pecos region of Texas they cause a neurological disease known as 'guajillo wobbles' or 'limerleg' which affects the animal's gait [3].

While isolating NMT from A. schweinfurthii, one of us (C.S.E.) developed acute migraine-type headaches, which